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Entropy Wikipedia

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Liquid Wikipedia

February 16th, 2019 - Liquid is a state of matter. In a liquid, the particles are close together but can move around. It is between a solid and a gas. The particles in a liquid have more energy than in a solid but less than in a gas. They are not fixed in place but can slide past each other. This allows the liquid to flow and take the shape of its container. The particles are still touching each other but not as tightly as in a solid. The forces between them are weaker than in a solid but stronger than in a gas. This is why a liquid has a definite volume but no definite shape. The particles in a liquid are in constant motion. They are vibrating and moving in all directions. The average kinetic energy of the particles is related to the temperature of the liquid. As the temperature increases, the particles move faster and the liquid expands. At a certain temperature, the particles have enough energy to break the forces that hold them together and the liquid turns into a gas. This is called boiling. The temperature at which this happens is called the boiling point. The boiling point of a liquid depends on the pressure. At higher pressures, the boiling point is higher. At lower pressures, the boiling point is lower. The boiling point of water at sea level is 100 degrees Celsius. At higher altitudes, the boiling point is lower. This is why it takes longer to cook food at high altitudes. The boiling point of a liquid also depends on the nature of the liquid. Some liquids have higher boiling points than others. For example, the boiling point of mercury is 357 degrees Celsius, while the boiling point of ethanol is 78 degrees Celsius. The boiling point of a liquid is a characteristic property of that liquid. It can be used to identify a liquid. The boiling point of a liquid is also important in many industrial processes. For example, in the distillation of crude oil, the different components are separated based on their boiling points. The boiling point of a liquid is also important in the design of heat exchangers and other equipment. The boiling point of a liquid is a key property that determines its behavior and its uses.

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